

The background is a light blue gradient. It is decorated with several realistic water droplets of various sizes. Some droplets are at the top left, some are at the bottom right, and others are scattered in the center. They have highlights and shadows, giving them a 3D appearance.

# DIALYSIS WATER AND DIALYSATE

DR. PARIN HEDAYATI

# INTRODUCTION

❖ 120-200 LITRE

❖ **CHEMICAL & MICROBIOLOGICAL PURITY OF DIALYSIS SOLUTION**

❖ DIALYSIS SOLUTION : PURIFIED WATER  
CONCENTRATE

# CONCENTRATES

❖ ELECTROLYTES

❖ COMMERCIAL SOURCES

4920ml

Acidic Hemodialysis Concentrated II  
solution

۴۹۲۰ میلی لیتر

همو دیالیز اسیدی غلیظ II  
محلول

## Instruction for use:

Only administrate under physician's order.  
Control acid & base concentration.  
The solution must be diluted immediately  
before use. To dilute 35 times, add 1 liter  
concentrated to 34 liters purified water  
and mix. Add sodium bicarbonate  
immediately before use. Do not use,  
if solution is not clear or contain particle.

## Warning:

The volume taken for use is to be measured  
accurately. Discard remaining solution  
after use. Keep in dry place and Between 4-30°C  
and protect from light. Use immediately after  
opening.

Keep out of reach and sight of children.

هشدار:  
- قبل از استفاده از محلول دستور العمل را به دقت مطالعه نمایید.  
- مخزوش بدون نسخه پزشک ممنوع است.  
- محلول باید بلافاصله پس از رقیق شدن استفاده گردد.  
- جهت ۳۵ بار رقیق نمودن محلول، ۱ لیتر محلول غلیظ را به ۳۴ لیتر آب خالص  
اضافه و مخلوط نمایید.  
- صحتیب بی کریستال باید بلافاصله قبل از مصرف به محلول اضافه گردد.  
- چنانچه محلول کدر یا دارای ذرات جامد قبل رقیق بود از مصرف آن خودداری کنید.  
- بلافاصله محلول را پس از استفاده دور بریزید.  
- دور از دسترس و دید اطفال نگهداری شود.  
- در ۴ تا ۳۰ درجه سانتیگراد نگهداری شود.

GTN 06260723880370  
UID 15746063703829071147  
LOT DL-970473  
EXP 2020/11/11

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Zip code: 33651-66596

PRICE: 204960 RIL

## Composition

Formula	Concentrated solution (g/l)		Diluted solution (35 times)(g/l)	
Sodium chloride	216.812		6.194	
Potassium chloride	5.218		0.149	
Calcium chloride,2H <sub>2</sub> O	6.431		0.183	
Magnesium chloride,6H <sub>2</sub> O	3.557		0.101	
Acetic acid (glacial)	7.356		0.210	
Dextrose,H <sub>2</sub> O	70.000		2.000	
<b>Electrolytes</b>	<b>mmol/l</b>	<b>mEq/l</b>	<b>mmol/l</b>	<b>mEq/l</b>
Na+	3710	3710	106.00	106.00
K+	70	70	2.00	2.00
Ca++	43.75	87.50	1.25	2.50
Mg++	17.50	35	0.50	1.00
CH <sub>3</sub> COO -	122.50	122.50	3.50	3.50
Cl -	3902.50	3902.50	111.50	111.50

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35X

# WATER CONTAMINANTS

❖ ALUMINIUM

❖ CHLORAMINE

❖ FLOURIDE

❖ COPPER

❖ ZINC

❖ LEAD

❖ BACTERIA & ENDOTOXINS

❖ TOXINS FROM BLUE –GREEN  
ALGAE

# ALUMINIUM

- FLOCCULATING AGENT
- AL SULFATE
- BONE DX , ANEMIA , ENCEPHALOPATHY SYNDROME

# CHLORAMINE

- ❖ PREVENTION OF BACTERIAL PROLIFERATION
- ❖ HEMOLYTIC ANEMIA

# **FLOURIDE**

- REDUCTION OF TOOTH DECAY
- SEVERE PRURITUS , NAUSEA , VF



# **COPPER & ZINC**

- LEACHING FROM METAL PIPES
- HEMOLYTIC ANEMIA

# **BACTERIA & ENDOTOXINS**

# **TOXINS FROM ALGAE**

- **TOXIC FOR HEMODIALYSIS PATIENTS**

# Water contaminants harmful for dialysis patient

## **Sign or Symptom Possible Water Contamination-Related Cause**

- Anemia - Aluminum, chloramines, copper, zinc
- Bone Disease - Aluminum, fluoride
- Hemolysis - Aluminum, copper, nitrates, chloramines
- Hypertension - Calcium, sodium
- Hypotension - Bacteria, endotoxins, nitrates
- Metabolic Acidosis - Low pH, sulfates
- Muscle Weakness - Calcium, Magnesium
- Nausea and Vomiting - Bacteria, calcium, copper, endotoxin, low pH, magnesium, nitrates, sulfates, zinc
- Neurological deterioration - Aluminum
- Fever, chills - Bacteria, endotoxin, copper, zinc

# FLUIDE QUALITY STANDARDS

ردیف	نوع آزمایش	واحد	حداکثر مجاز	نتیجه آزمایش
۱	کدورت	NTU	۵	0.42
۲	pH	pH	۶/۵-۹	7.5
۳	کلر باقیمانده	میلی گرم در لیتر	۰/۲-۰/۸	-
۴	هدایت الکتریکی	میکروزیمنس بر سانتیمتر	-	422
۵	کل مواد جامد محلول	میلی گرم در لیتر	۱۵۰۰	295
۶	شوری	p.p.t	-	0.1
۷	سختی کل	میلی گرم در لیتر $\text{CaCO}_3$	۵۰۰	202
۸	قابلیت کل	میلی گرم در لیتر $\text{CaCO}_3$	-	186
۹	کربنات	میلی گرم در لیتر $\text{CaCO}_3$	-	0
۱۰	بی‌کربنات	میلی گرم در لیتر $\text{CaCO}_3$	-	226.92
۱۱	کلراید	میلی گرم در لیتر $\text{Cl}^-$	۲۰۰	23
۱۲	سولفات	میلی گرم در لیتر $\text{SO}_4^{--}$	۲۰۰	36.4
۱۳	فلوراید	میلی گرم در لیتر $\text{F}^-$	۰/۵-۱/۵	0.19
۱۴	نیترات	میلی گرم در لیتر $\text{NO}_3^-$	۵۰	8.2
۱۵	نیتریت	میلی گرم در لیتر $\text{NO}_2^-$	۳	0
۱۶	کلسیم	میلی گرم در لیتر $\text{Ca}^{++}$	۲۰۰	49.64
۱۷	منیزیم	میلی گرم در لیتر $\text{Mg}^{++}$	۲۰	19.17
۱۸	سدیم	میلی گرم در لیتر $\text{Na}^+$	۲۰۰	14.8
۱۹	پتاسیم	میلی گرم در لیتر $\text{K}^+$	-	0.5

<b>Contaminant</b>	<b>Maximum Concentration mg/L</b>	<b>Test Methodology</b>
<b>Calcium</b>	2 (0.1 mEq/L)	EDTA or Atomic Absorption
<b>Magnesium</b>	4 (0.3 mEq/L)	Atomic Absorption
<b>Potassium</b>	8 (0.2 mEq/L)	Atomic Absorption, or Flame Photometry
<b>Sodium</b>	70 (3.0 mEq/L)	Atomic Absorption or Flame Photometric
<b>Antimony</b>	0.006	Atomic Absorption (platform)
<b>Arsenic</b>	0.005	Atomic Absorption (gaseous hydride)
<b>Barium</b>	0.10	Atomic Absorption (electrothermal)
<b>Beryllium</b>	0.0004	Atomic Absorption (platform)
<b>Cadmium</b>	0.001	Atomic Absorption (electrothermal)
<b>Chromium</b>	0.014	Atomic Absorption (electrothermal)
<b>Lead</b>	0.005	Atomic Absorption (electrothermal)
<b>Mercury</b>	0.0002	Flameless Cold Vapor (Atomic Absorption)

<b>Contaminant</b>	<b>Maximum Concentration mg/L</b>	<b>Test Methodology</b>
<b>Selenium</b>	0.09	Atomic Absorption (gaseous; or electrothermal)
<b>Silver</b>	0.005	Atomic Absorption (electrothermal)
<b>Aluminum</b>	0.01	Atomic Absorption (electrothermal)
<b>Chloramines</b>	0.10	DPD Ferrous Titrimetric Method
<b>Total chlorine</b>	0.50	DPD Ferrous Titrimetric Method
<b>Copper</b>	0.10	Atomic Absorption (direct aspiration)
<b>Fluoride</b>	0.20	Ion Selective Electrode Method
<b>Nitrate (as N)</b>	2.00	Cadmium Reduction Method
<b>Sulfate</b>	100.00	Turbidimetric Method
<b>Thallium</b>	0.002	Atomic Absorption (platform)
<b>Zinc</b>	0.10	Atomic Absorption (direct aspiration)

Association for the Advancement of Medical Instrumentation. *Dialysate for hemodialysis* (ANSI/AAMI RD52:2004). Arlington (VA). American National Standard. 2004



# FLUID QUALITY STANDARDS

## PRODUCT WATER

- ❖ BACTERIA  $< 100$  CFU/ML
- ❖ ENDOTOXIN  $< 0.25$  EU/ML

# FLUID QUALITY STANDARDS

## FINAL DIALYSIS SOLUTION

- ❖ BACTERIA  $\leq 100$  CFU/ML
- ❖ ENDOTOXIN  $\leq 0.5$  EU/ML



**ULTRAPURE DIALYSIS SOLUTION**

**CHRONIC INFLAMMATION**



# ULTRAPURE DIALYSIS SOLUTION

❖ **BACTERIA < 0.1 CFU/ML**

❖ **ENDOTOXIN < 0.03 EU/ML**

# ULTRAPURE DIALYSIS SOLUTION

❖ CRP

❖ B2MICROGLOBULIN

❖ IL-6

❖ RRF

❖ IMPROVED RESPONSE TO EPO

❖ CV MORBIDITY

❖ BETTER NUTRITION      ALBUMIN

MIDARM CIRCUMSTANCE

# **METHODS OF PURIFYING WATER FOR HD**

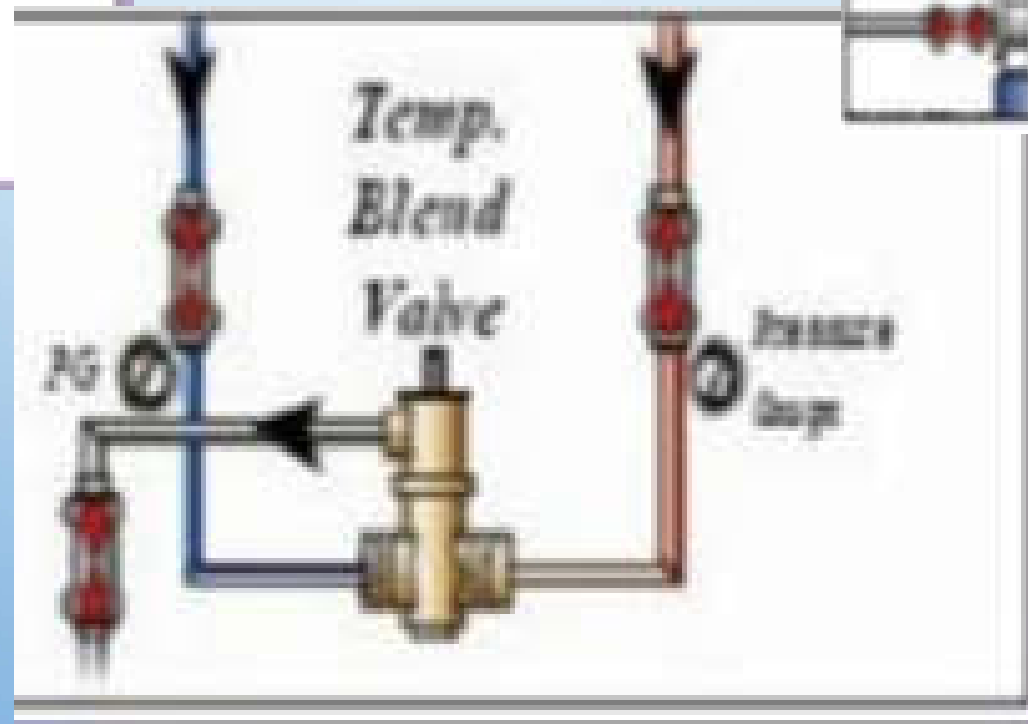
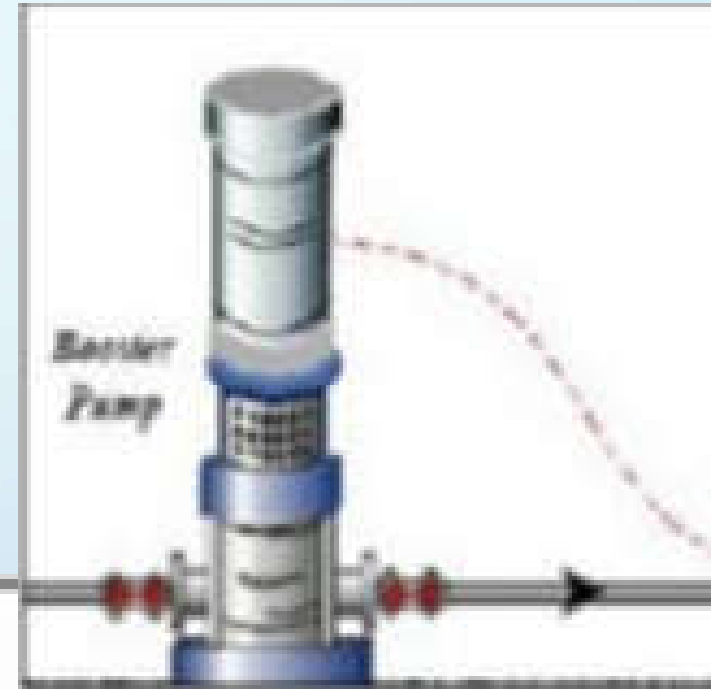
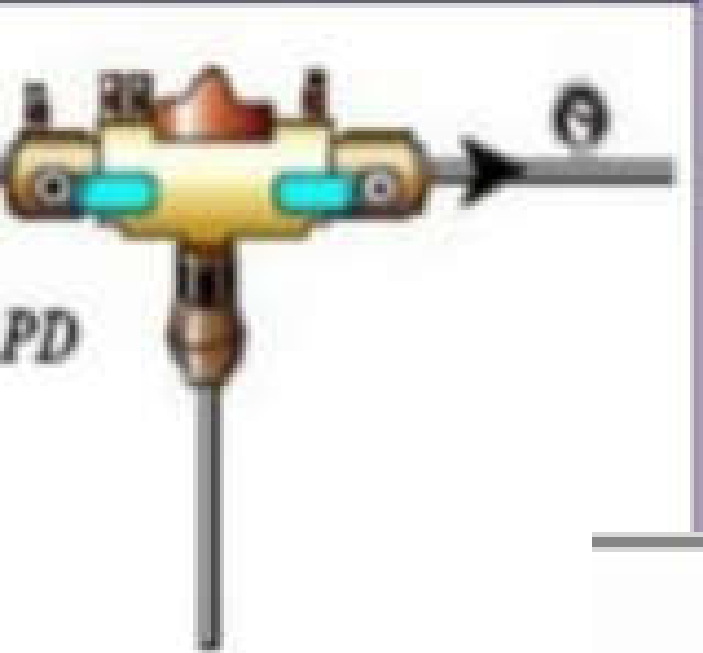
❖ **PRETREATMENT**

❖ **PRIMARY PURIFICATION**

❖ **DISTRIBUTION TO THE POINT OF USE**

# **PRETREATMENT**

- ❖ **BLENDING HOT AND COLD WATER**
- ❖ **WATER SOFTENING**
- ❖ **FILTRATION THROUGH ACTIVATED CARBON**
- ❖ **CORRECTION OF PH**



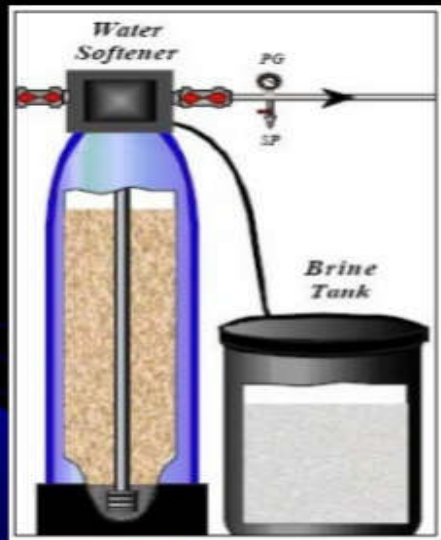


# **WATER SOFTENER**

- ❖ REMOVING CA AND MG BY EXCHANGE FOR NA  
BOUND IONICALLY TO A RESIN BED
- ❖ BACKWASHING AND REGENERATION OF RESINS

# WATER SOFTNER

## Water Softener



# **CARBON BEDS**

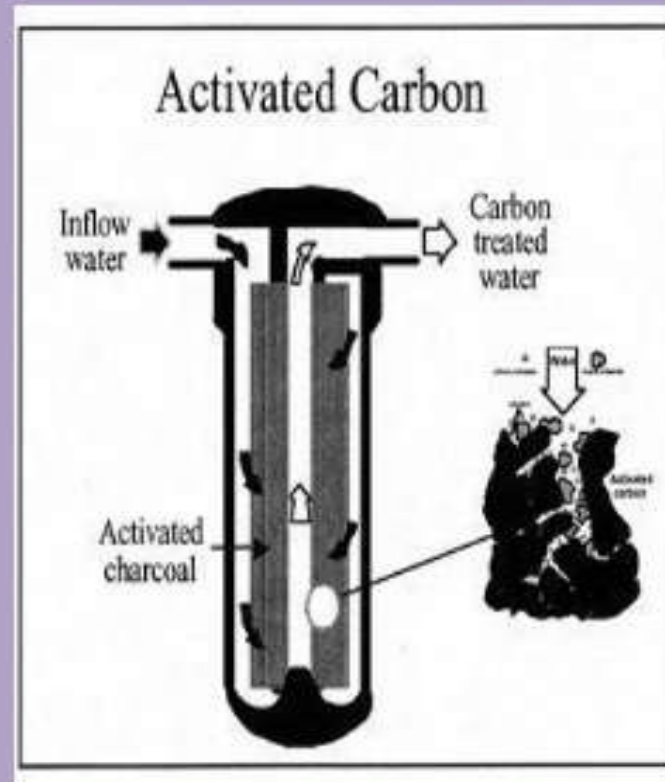
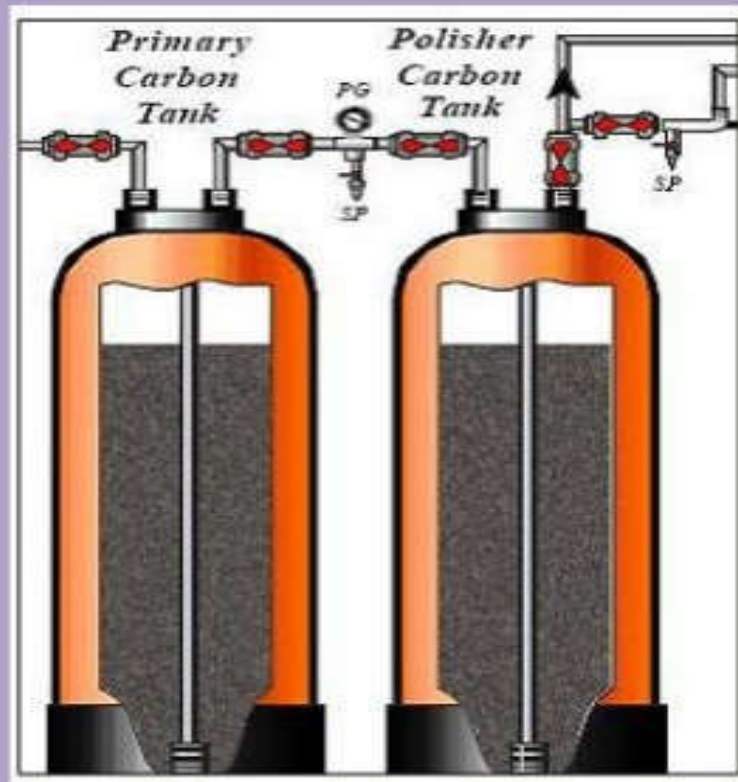
- ❖ REMOVING CHLORINE AND CHLORAMINE
- ❖ ORGANIC COMPOUNDS

• CHLORINE + ORGANIC COMPOUNDS →

CANCER CAUSING COMPOUNDS

■ CHLORINE → CHLORAMINE

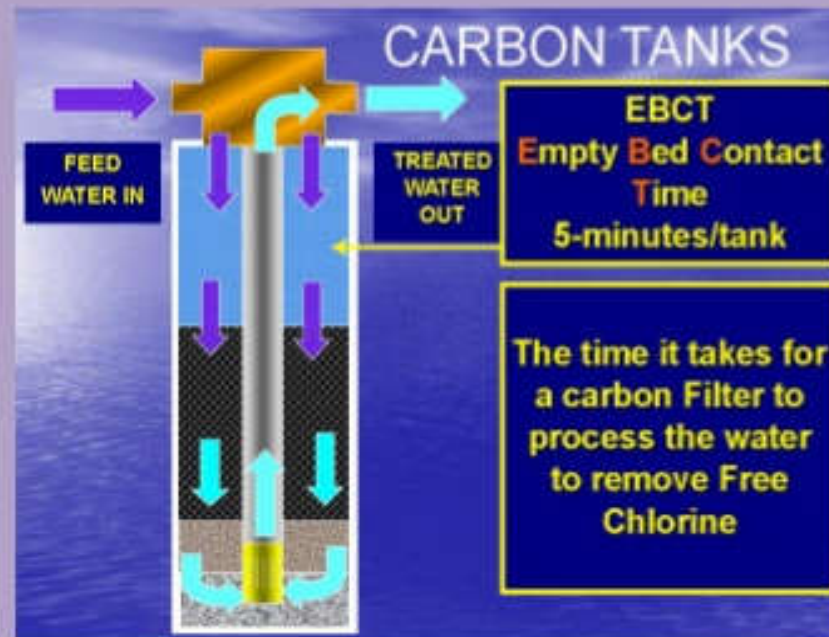
■ CHLORAMINE : HEMOLYTIC ANEMIA



# CARBON BEDS

## Carbon Tank

- Removes chlorine and chloramine
- These are high level oxidative chemicals. They are added to municipal water systems to kill bacteria, but they also cause hemolysis



# **CARBON BEDS**

❖ **TOTAL CHLORINE LEVEL**

# CRITICAL ASPECT TO PROPER FUNCTIONING OF CARBON BEDS

- ❖ CONTACT TIME      10 MIN
- ❖ REGULAR BACKWASHING OF CARBON BEDS
- ❖ ADJUSTMENT OF THE PH OF THE FEED WATER
- ❖ CORROSION INHIBITORS



# **PRIMARY PURIFICATION PROCESS**

❖ **REVERSE OSMOSIS**

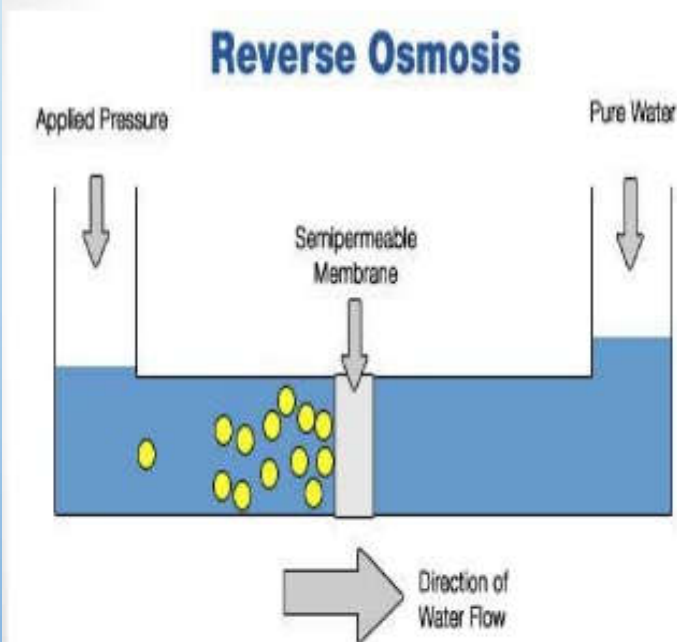
❖ **DEIONIZATION**

# REVERSE OSMOSIS

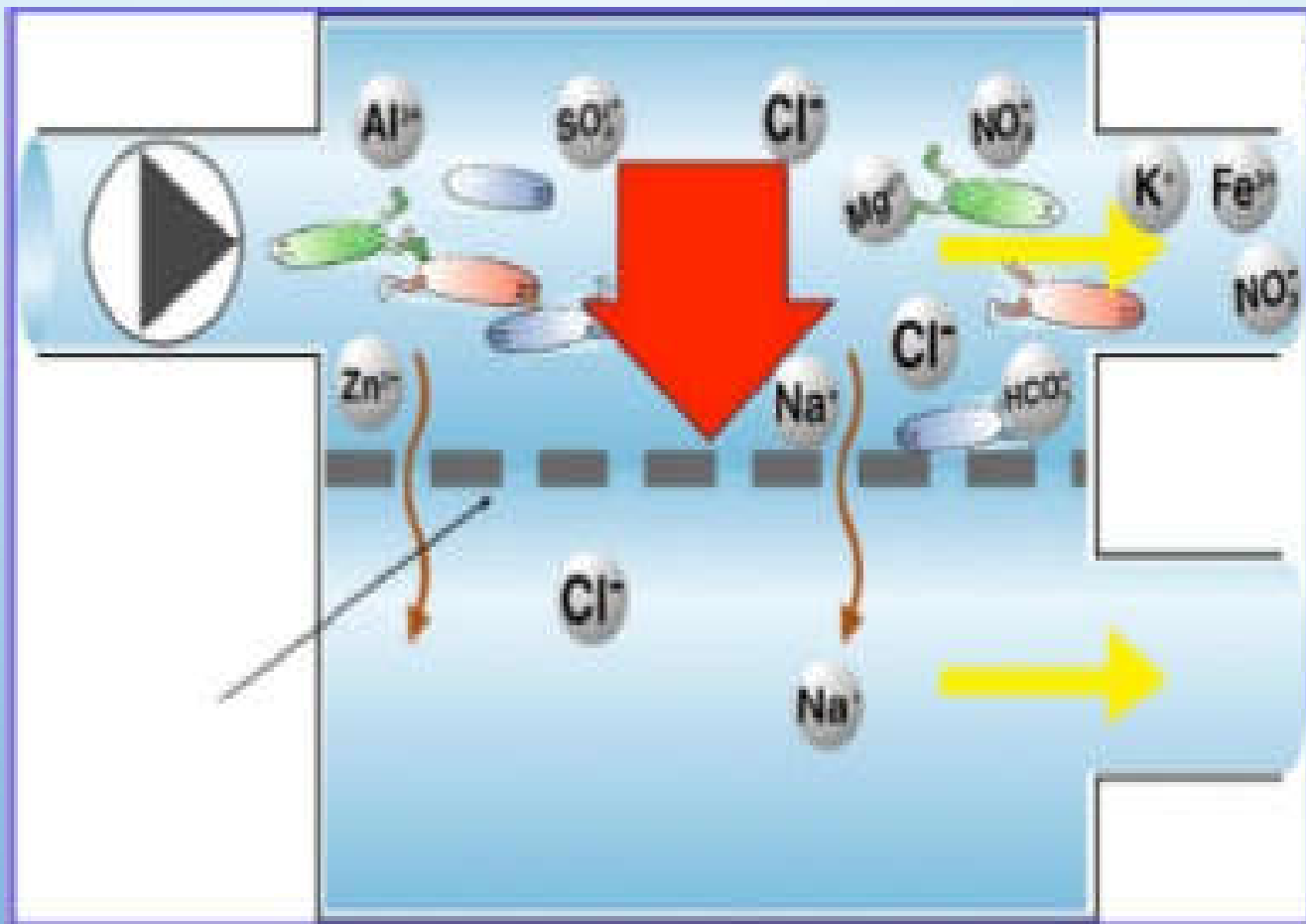


# REVERSE OSMOSIS

## Primary purification process



# REVERSE OSMOSIS



# REVERSE OSMOSIS


- ❖ HIGH PRESSURE FILTRATION OF WATER THROUGH A SEMIPERMEABLE MEMBRANE
- ❖ DISSOLVED SOLUTES
- ❖  $> 95\%$
- ❖ EFFECTIVE BARRIER AGAINST BACTERIA AND ENDOTOXINS

# DEIONIZATION

- ❖ DO NOT REMOVE NONIONIC CONTAMINANTS ,  
BACTERIA , ENDOTOXINS
- ❖ BOTH CATIONS AND ANIONS



# DEIONIZATION

## ❖ CATIONIC RESIN :

SULFURIC GROUPS :  $H^+$    $Na, Ca, Al.$

## ❖ ANIONIC RESIN : AMMONIUM

$OH^-$    $CHLORIDE, PHOSPHATE, FLUORIDE.$

❖  $H^+$    $OH^-$    $H_2O$

# DEIONIZATION

## • MONITORING

CONDUCTIVITY OF OUTFLOW WATER



# DEIONIZATION

- LARGE SURFACE AREA FOR BACTERIAL PROLIFERATION

- ❖ **ULTRAFILTER**

- ❖ **UV LIGHT** : LIPOPOLYSACCHARIDE & PEPTIDOGLYCAN

# **DISTRIBUTION OF PURIFIED WATER**

- INERT MATERIAL : CHEMICAL CONTAMINANTS
- MICROBIAL CONTAMINANTS :
- APPROPRIATE DESIGNED PIPING SYSTEMS

**DISINFECTION**

# **DISTRIBUTION OF PURIFIED WATER**

❖ **LOOP**

❖ **NO MULTIPLE BANCHING AND DEAD ENDS**

❖ **STORAGE TANKS : MINIMUM SIZE , TIGHT LID , EASE  
OF DISINFECTION**

# **DISTRIBUTION OF PURIFIED WATER**

## **❖ REGULAR DISINFECTION**

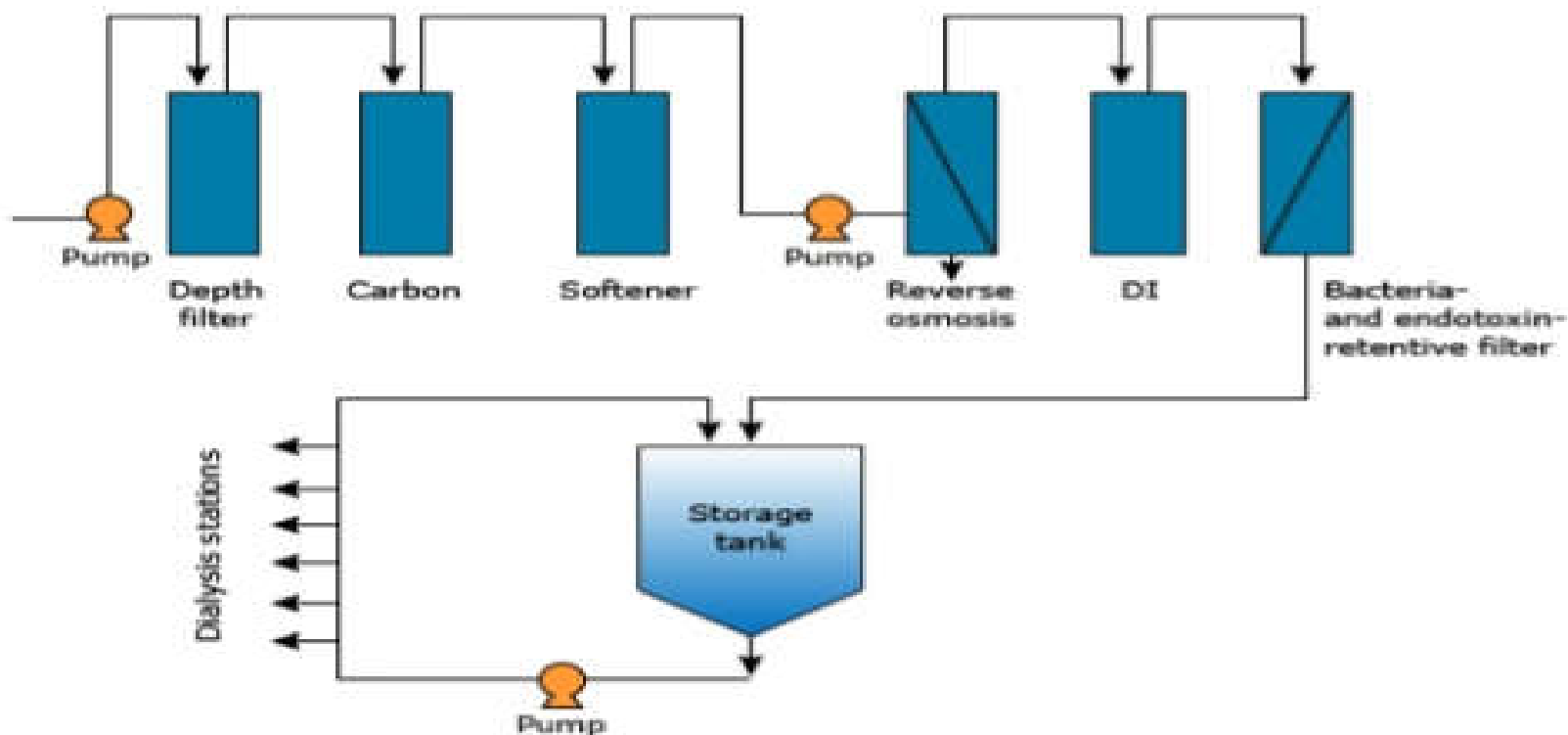
**CHEMICAL GERMICIDE : MONTHLY**

**HOT WATER OR OZONE**

- **SOLUTION CULTURE**

- **ENDOTOXIN TESTS**

# Schematic diagram example of a water treatment system for hemodialysis



# Purification Processes

Process	Contaminant
Carbon Adsorption	Chloramine, organics
Softener	Calcium, Mg
Reverse osmosis	Ionic contaminants, bacteria, endotoxin
Deionization	Ionic contaminants
Ultrafiltration	Bacteria, endotoxin

## Testing of samples

Testing of water samples shall be carried out by **trained and accredited persons** or accredited laboratories.

The dialysis unit **shall maintain records of persons** who have been trained and accredited and full details of accredited laboratories.

The records shall be maintained within the dialysis unit.

# Sample collection

## Water sample sites

Samples are to be taken at **outlets of the water distribution system**.

Prior to sampling, the inside of the outlet can be disinfected, especially if no hemodialysis machine is attached. The reason for such disinfection is that, over time, residual water in an outlet will support microbial growth. The disinfection can be made by flushing the inside of the **outlet with 70 % ethanol** or iso-propanol. A sterile cotton swab wetted with alcohol can also be used. Exposure time is to be >15 s.

It is sufficient to let out enough water to **rinse off the alcohol** (200 ml to 500 ml) prior to sampling.

Alternatively, hoses can be disconnected from the tap and the taps opened and allowed to flush for 2 min to 3 min before aseptically collecting a sample.

Sample for **cultivation and endotoxin analysis**: Sample volume 5 ml to 1,000 ml or volume as specified by the laboratory..



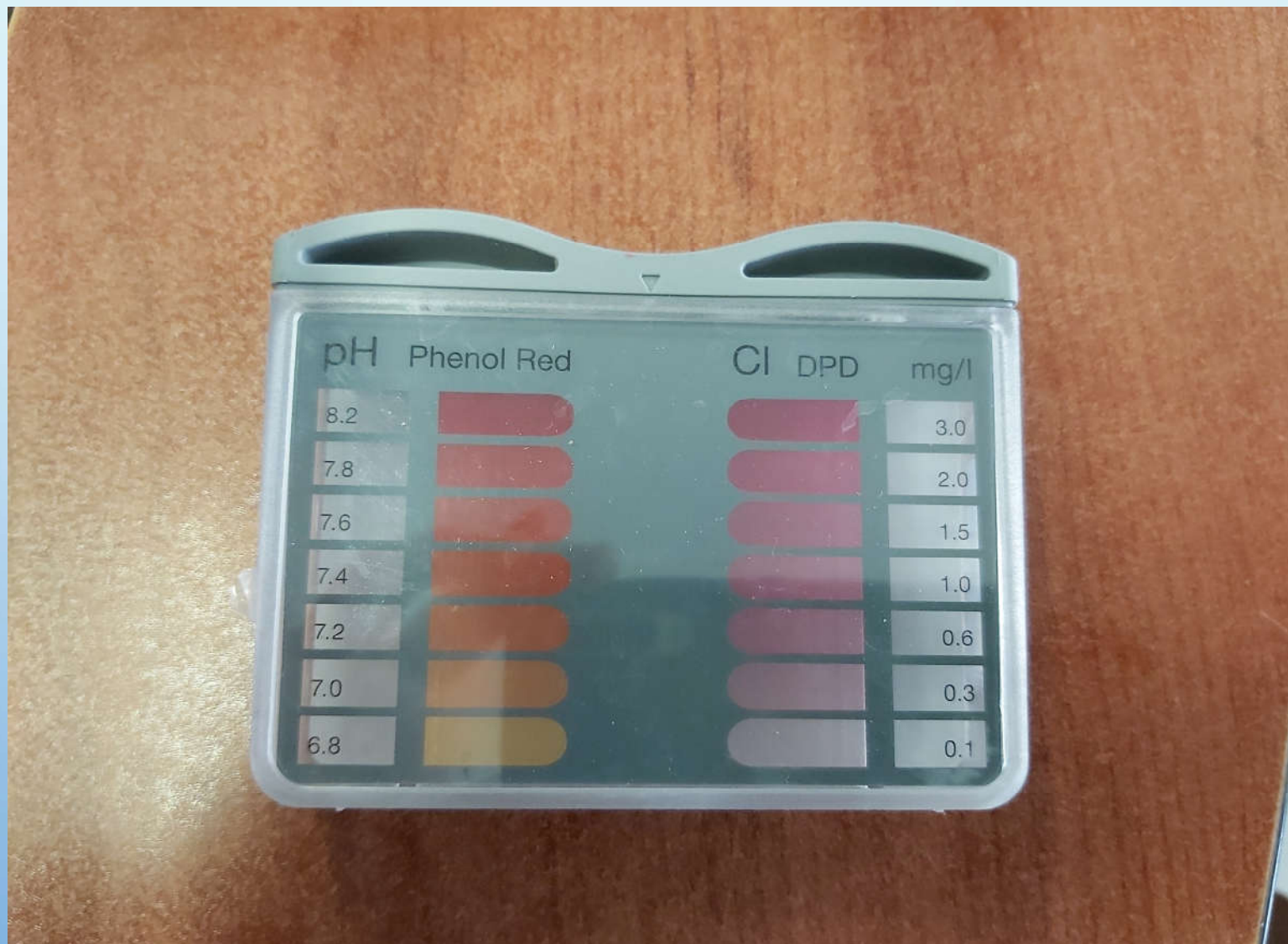
# Storage of samples

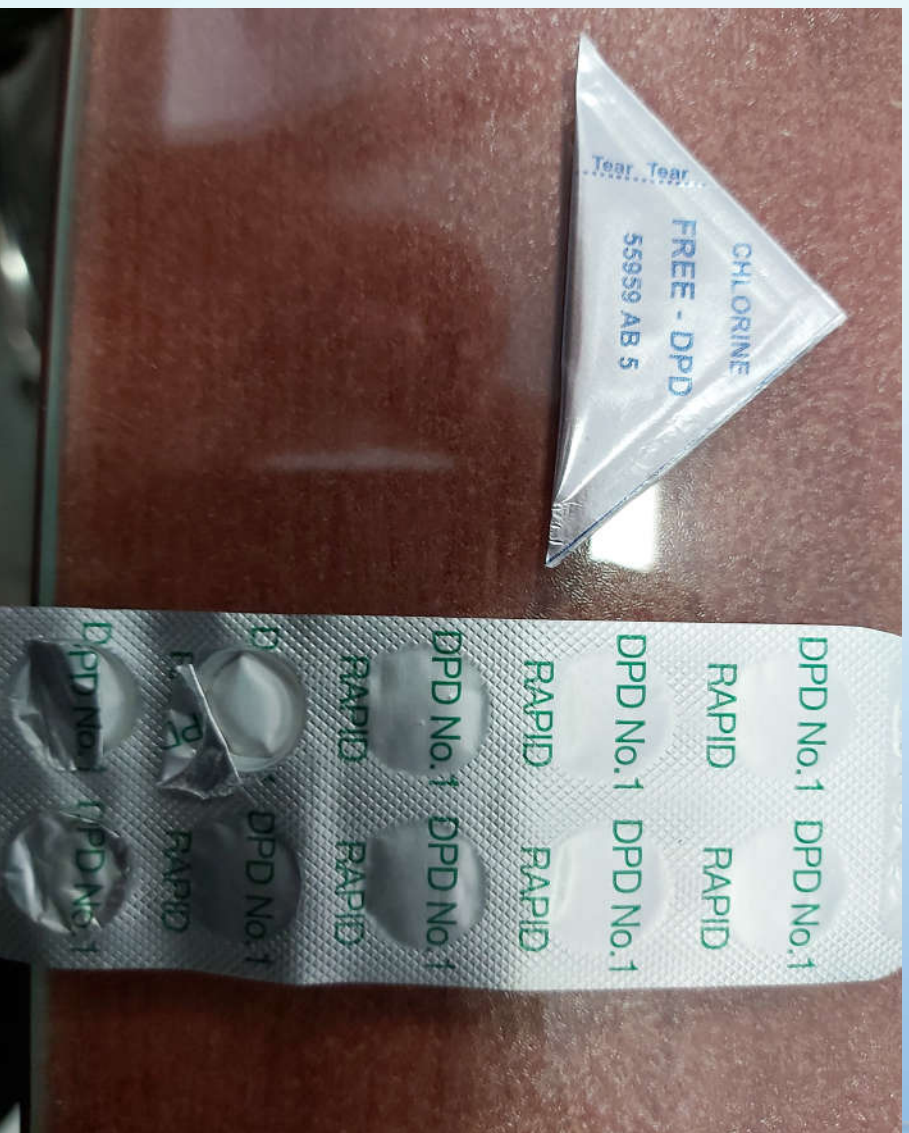
Heterotrophic plate count

## Storage of samples

Microbial analysis of water and dialysis fluid samples **should be conducted as soon as possible** after collection to avoid unpredictable changes in the microbial population. If samples cannot be analyzed within 4 h of collection, **follow the laboratory's instructions** for shipping. Samples intended for colony counts should not be frozen.

Storage of samples for **endotoxin** analysis may be different from what is given above, provided the complete procedure **follows the manufacturer's instructions** for use of the LAL assay.









## Water treatment unit

TDS, and conductivity **daily**

Microbiological culture and endotoxin assay.  
**Monthly**

Chemical assay **every 6 months**

# Water treatment unit

- There is no limit for RO product TDS/conductivity.
- Values that are acceptable in one location may not be acceptable in another location.
- TDS in some areas of 50 ppm is acceptable and other areas where **10 ppm is NOT acceptable**. It all depends on your raw water.
- A slight change in the amount **of Fluoride** injected into the water can cause the RO product water to go less than a 1 ppm TDS increase.

# **DIALYSIS SOLUTION PREPARATION**

## **❖ PROPORTIONING MACHINES**

**FIXED VOLUMES OF DIALYSATE CONCENTRATE WITH  
FIXED VOLUME OF WATER**

**CONDUCTIVITY BASED SYSTEMS**

# DIALYSIS SOLUTION PREPARATION

❖ DUAL CONCENTRATE SYSTEM

BICARBONATE : PH 8.0

CA AND MG PRECIPITATION

❖ **2 CONCENTRATES** : BICARBONATE

ACID( CITRIC ACID OR ACETIC ACID)





# **DIALYSIS SOLUTION PREPARATION**

## **BICARBONATE LEVEL ON THE MONITOR**



**4920ml**  
**Acidic Hemodialysis Concentrated II**  
**solution**



۴۹۲۰ میلی لیتر  
 همو دیالیزاسیدی غلیظ II  
 محلول

**Instruction for use:**

Only administrate under physician's order.  
 Control acid & base concentration.  
 The solution must be diluted immediately  
 before use. To dilute 35 times, add 1 liter  
 concentrated to 34 liters purified water  
 and mix. Add sodium bicarbonate  
 immediately before use. Do not use,  
 if solution is not clear or contain particle.

**Warning:**

The volume taken for use is to be measured  
 accurately. Discard remaining solution  
 after use. Keep in dry place and Between 4-30°C  
 and protect from light. Use immediately after  
 opening.

**Keep out of reach and sight of children.**

هشدار:  
 - قبل از استفاده از محلول دستور العمل را به دقت مطالعه نمایید.  
 - فروش بدون نسخه پزشک ممنوع است.  
 - محلول باید بلافاصله پس از رقیق شدن استفاده گردد.  
 - جهت ۳۵ بار رقیق نمودن محلول، ۱ لیتر محلول غلیظ را به ۳۴ لیتر آب خالص  
 اضافه و مخلوط نمایید.  
 - سدیم بی کربنات باید بلافاصله قبل از مصرف به محلول اضافه گردد.  
 - چنانچه محلول کدر یا دارای ذرات جامد قبل رقیق بود از مصرف آن خوداری کنید.  
 - باقیمانده محلول را پس از استفاده دور بریزید.  
 - دور از دسترس و دید اطفال نگهداری شود.  
 - دور از نور مستقیم آفتاب نگهداری شود.

**Composition**

Formula	Concentrated solution (g/l)		Diluted solution (35 times)(g/l)	
Sodium chloride	216.812		6.194	
Potassium chloride	5.218		0.149	
Calcium chloride,2H <sub>2</sub> O	6.431		0.183	
Magnesium chloride,6H <sub>2</sub> O	3.557		0.101	
Acetic acid (glacial)	7.356		0.210	
Dextrose,H <sub>2</sub> O	70.000		2.000	
Electrolytes	mmol/l	mEq/l	mmol/l	mEq/l
Na+	3710	3710	106.00	106.00
K+	70	70	2.00	2.00
Ca++	43.75	87.50	1.25	2.50
Mg++	17.50	35	0.50	1.00
CH3COO -	122.50	122.50	3.50	3.50
Cl -	3902.50	3902.50	111.50	111.50

GTN 06260723880370  
 UID 15746063703829071147  
 LOT DL-970473  
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# FLUID DIALYSIS SOLUTION COMPOSITION

- NA      SODIUM PROFILING
- K
- CA
- BICARBONATE

# **FLUID DIALYSIS SOLUTION COMPOSITION**

# DIALYSIS SOLUTION **CALCIUM** LEVELS

❖ **ACUTE DIALYSIS** : 3-3.5 MEQ/L

❖  $CA < 3$  → HYPOTENSION DURING DIALYSIS

❖ PREDIALYSIS HYPOCALCEMIA : ACIDOSIS CORRECTION →  
LOWERING CA LEVEL → SEIZURE

↑  
QT DISPERSION

# DIALYSIS SOLUTION CALCIUM LEVELS

- CHRONIC DIALYSIS : 2.5 MEQ/L

# **DIALYSIS SOLUTION CALCIUM LEVELS**

- DIALYTIC TREATMENT OF ACUTE HYPERCALCEMIA
  - ADDING 2.5 MEQ/L CALCIUM TO THE HEMODIALYSIS SOLUTION TO PREVENT RAPID DECREASE IN IONIZED CALCIUM LEVEL
  - ❖ FREQUENT MEASUREMENT AND P/CA DURING DIALYSIS

# DIALYSIS SOLUTION **POTASSIUM** LEVEL

❖ ACUTE DIALYSIS : 2-4.5 MEQ/L

❖ PREDIALYSIS K < 4.5 MEQ/L → DIALYSIS SOLUTION K  $\geq$  4 MEQ/L

❖ PREDIALYSIS K > 5.5 MEQ/L → DIALYSIS SOLUTION K 2 MEQ/L

❖ AT RISK OF ARRHYTHMIA OR DIGOXIN → DIALYSIS SOLUTION K 2.5-3.5 MEQ/L

❖ PREDIALYSIS K > 7 MEQ/L → DIALYSIS SOLUTION K < 2 MEQ/L



# DIALYSIS SOLUTION **POTASSIUM** LEVEL

• CHRONIC DIALYSIS : 2MQ/L

• EXCEPTION : USUAL PREDIALYSIS PLASMA K < 4.5 MQ/L(MALNOURISHED PATIENTS)

DIGOXIN

: DIALYSIS K 3 MQ/L

BUT : INTERDIALYSIS PLASMA K MEASUREMENT : ? KAYEXALATE

# DIALYSIS SOLUTION **SODIUM** LEVEL

• **ACUTE DIALYSIS : 145 MQ/L**

• HYPONATREMIA : PREDIALYSIS NA < 130

MINIMUM NA & SLOW BFR(50-100) & NO LONGER THAN 1 HOUR & CHECK NA 30-60 MIN

OR DELAY DIALYSIS

• HYPONATREMIA : PREDIALYSIS NA > 130

NA:< 10 MQ/L ABOVE SERUM NA

NA: 140-145MQ/L

# DIALYSIS SOLUTION **SODIUM** LEVEL

- HYPERNATREMIA
- DIALYSIS : DANGEROUS
- MAXIMUM : 3-5 MQ/L

# DIALYSIS SOLUTION **SODIUM** LEVEL

- CHRONIC DIALYSIS :
- 135-145 MEQ/L
- > 138 : THIRST
- < 135: HYPOTENSION

# NA PROFILING

- THE SODIUM CONCENTRATION IN THE DIALYSIS FLUID, INSTEAD OF BEING CONSTANT, FOLLOWS A TIME-DEPENDENT PROFILE OVER THE COURSE OF A HEMODIALYSIS SESSION.
- AVOID OSMOTIC DISEQUILIBRIUM BY KEEPING PLASMA OSMOLALITY IN THE PHYSIOLOGICAL RANGE.
- REDUCTION IN THE INCIDENCE OF MUSCLE CRAMPS, IMPROVED SODIUM REMOVAL, AND IMPROVED VASCULAR STABILITY.

# NA PROFILING

- THERE CAN BE NEGATIVE CONSEQUENCES TO A HIGH DIALYSATE SODIUM CONCENTRATION.
- SODIUM CAN ACCUMULATE IN THE PATIENT : INCREASED POST-DIALYSIS THIRST, INCREASED INTERDIALYTIC WEIGHT GAIN AND THE DEVELOPMENT OF HYPERTENSION.
- SVS WAS DEVELOPED TO ACHIEVE THE BENEFITS OF HIGH PLASMA SODIUM LEVELS WHILE AT THE SAME TIME AVOIDING UNNECESSARY HIGH INTRADIALYTIC SODIUM UPTAKE BY THE PATIENT .

# DIALYSIS SOLUTION **BICARBONATE** LEVEL

## • ACUTE DIALYSIS

• DIALYSIS MACHINES : FIXED : BICARBONATE : 32 MEQ/L

CHANGEABLE : BICARBONATE : 20 MEQ/L

❖ 4-8 MEQ/L

# DIALYSIS SOLUTION **BICARBONATE** LEVEL

## • ACIDOSIS :

METABOLIC : BICARBONATE < 10MEQ/L : 20-25 ( SERUM BICARBONATE : 15-20)

RESPIRATORY : HIGHER BICARBONATE RANGE



# DIALYSIS SOLUTION **BICARBONATE** LEVEL

• ACUTE DIALYSIS :

• ALKALOSIS:

METABOLIC: SERUM BICARBONATE  $>28$  : DIALYSIS BICARBONATE : LOW : 20-28

RESPIRATORY : DIALYSIS BICARBONATE : 15-20

# DIALYSIS SOLUTION **BICARBONATE** LEVEL

• CHRONIC DIALYSIS :

• GOAL : SERUM BICARBONATE : 20-23

ACID : ACETATE : BICARBONATE : 32 + 4

CITRATE : BICARBONATE : 28+ 8

# **Water treatment for HD Unit**

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THANK YOU